B. E. (Met) 7th Semester Final Examination 2012

Nanostructured and Functionally graded materials. (MT – 705/1)

Full marks: 70 Time 3hrs

(Use separate answerscriptfor different sections)

SECTION A

(Answer any five questions)

Q1. Explain why HRTEM images of nanostructured materials should be interpreted carefully.	(=)
	(7)
Q2. Explain why the specific heat capacity and thermal conductivity of nanocrystalline materials is different from their coarse grained counterparts.	3
	(7)
Q3. Explain how the crystal sizes of nanocrystalline ZnO affect its luminescence property.	()
Property.	(7)
Q4. Explain why the chemical binding effects at the grain boundaries of diamond like structures changes with decreasing grain size.	
or not made of the control of the co	(7)
Q5. With reference to the rapid expansion of supercritical fluid solution, explain how hydro-thermal synthesis technique is used to prepare nanocrystaline materials.	
	(7)
Q6. State and explain the conditions required for preparing a super-lattice structure Cu/Ni is over a conducting substrate by employing electro-deposition technique.	e of
	(7)
Q7. State the use and importance of SPION's in the biomedical field.	
	(7)
SECTION B	

- Q7. Answer any two: $(71/2 \times 2)$
 - (i) Describe the favourable condition for glass forming ability of metallic alloys.
 - (ii) Describe the variation specific volume as the function of temperature for crystalline and amorphous alloys indicating the variation of Tg for different cooling rates.
 - (iii) Enumerate the conditions for appearance of glass transition in terms of a binary phase diagram comprising two terminal eutectics and an intermediate phase.
 - (iv) Enumerate the technological properties of glass in the light of its featureless microstructure

Q8.	Compare and contrast (any hie) (i) Glass and supercooled liquids (ii) Strong and fragile glass (iii) Atomic mobility in amorphous and crystalline alloys (iv) Plasticity in crystalline and glassy alloys	(5×2
Q9.	Write short notes (any two) (i) VTF equation (ii) Kauzmann's Paradox (iii) Submerged eutectic (iv) Reduced glass transition temperature	(5×2)
