B.E. (Met.), 3 rd Semester Final Examination, 2013

INTRODUCTION TO MATERIALS (MT – 301)

Time: 3 hrs Full Marks: 70

Answer Any Seven Questions

1. (a) Draw the following

1.5 x 4+4=10

- (i) (111) in cubic crystal
- (ii) [110] in cubic crystal
- (iii) (001) in tetragonal crystal
- (iv) (0001) in hexagonal closed packed plane
- (b) Discuss Bravais lattice and its varieties
- 2. Differentiate between (any five)

 $5 \times 2 = 10$

- a. Schottky and Frankel Imperfections
- b. Slip and Twinning
- c. Edge and Screw Dislocation
- d. Malleability and Ductility
- e. Stiffness and Toughness
- f. Allotrope and Polymorph
- 3. Write Short note on the following (any four)

 $4 \times 2.5 = 10$

- a. Hume-Rothery Rules for Mixing
- b. Slip systems in Aluminium Alloys
- c. Determination of Yield Point in a stress strain curve
- d. Classification of Engineering Materials
- e. Relation Between Lattice and Basis
- 4. Answer the following numerical problems (any four)

 $2.5 \times 4 = 10$

- a. Find the equilibrium vacancies in 1 m3 of Copper at 1000°C; Given: density of Copper is 8.4 g/cc, atomic weight of Copper is 63.5 g/mol, activation energy is 0.9 eV/atom
- b. Determine the density of BCC iron, which has a lattice parameter of 0.2866 nm.
- c. Calculate the c/a ratio for hcp crystal

- d. Calculate the planar atomic density on the (110) plane of the α iron BCC lattice in atoms per square millimeter. The lattice constant of α iron is 0.2787.
- e. Calculate the linear atomic density on the [110] direction in the Copper crystal lattice in atoms per millimeter. Copper has a lattice constant of 0.361 nm.
- 5. Justify the Following (any four)

- $2.5 \times 4 = 10$
- a. Stacking fault in fcc may locally create hcp stacking
- b. Atomic Packing Factor (APF) for fcc crystal structure is 0.74
- c. Metallic bonded solids more ductile than covalent bonded ones
- d. Phase diagram can give indications about equilibrium situations.
- e. Degree of freedom (DOF) is zero for invariant reactions.
- 6. (a) Write different reactions seen in Fe-Fe₃C phase diagram. 4+3+3=10
 - (b) Define Intermetallic Compound. How they are different from other compounds?
 - (c) Discuss the nature of Pearlitic transformation.
- 7. Describe with diagram the various types of flames that can be produced in Oxyacetylene flame welding. State the advantages, disadvantages and applications of Oxy-acetylene flame welding. 6 + 4 = 10
- 8. State the advantages of Rightward technique over Leftward. 3 + 7 = 10 Explain the role of Fluxes in Oxy-acetylene flame welding.
- 9. State the Factors for Selection of right kind of Electrode for a particular welded joint. State and explain the Functions of Flux Coating Ingredients. 3 + 7 = 10
- 10. Describe the process of manufacturing flux coated electrodes for MMAW. 10