B.E. (Met) Part IV 7th Semester Examination, 2011

DEGRADATION OF MATERIALS AND THEIR PROTECTION (MT 704)

Time: 3 hours Full marks 70

Answer any seven questions

All questions carry equal marks

- 1 (a) What is polarisation? Explain with suitable examples, how polarisation influences corrosion processes.
 - (b) What is exchange current density? Discuss its importance in corrosion.

[(5+5)]

- 2 (a) Differentiate between EMF series and Galvanic series. Why is latter more useful in the prediction of galvanic corrosion?
 - (b) Explain with examples, the area effect in galvanic corrosion.

[7+3]

- (a) Describe the phenomenon and mechanism of weld decay in austenitic stainless steels.
- (b) What is cavitation damage?

[8+2]

- (a) Discuss briefly the electrochemical mechanisms of stress corrosion cracking and observations in their support.
- (b) Discuss the phenomenon and cause of high temperature hydrogen attack in steels.

[5+5]

5 Explain why:

3

4

- (a) Even the stabilized stainless steels are also some time prone to inter-granular corrosion.
- (b) Copper does not corrode in air-free acids.
- (c) Oxidisers are some time referred to as dangerous inhibitors.
- (d) Sulphides are detrimental for steel for use in sour environment.

[22×4]

- 6 (a) Briefly discuss the mechanism of formation and growth of oxidation.
 - (b) Discuss the mechanism of p-type and n-type oxide film. Discuss the effect of oxidation rate on doping the lower and higher valance materials in the n-type and p-type oxide film.

[4+6]

- 7 (a) A 100 mm² cross section sample of 99.9 wt % nickel, 0.8mm thick, is oxidized in oxygen at 1atm. pressure at 600°C. After 2 hours, the sample shows a weight gain of 70μgms/cm². If this material shows parabolic oxidation behaviour, what will the weight gain be after 20hours?
 - (b) Discuss briefly the various types of oxidation rate laws with example.
 - (c) Mention various methods used for applying metallic coats.

[4+4+2]

ear.
[5+5]
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[3+4+3]